Acta Crystallographica Section E

## Structure Reports

Online
ISSN 1600-5368

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## Key indicators

Single-crystal X-ray study
$T=295 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.037$
$w R$ factor $=0.115$
Data-to-parameter ratio $=19.9$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
# Methyl $\beta$ - $N$-phenylmethylenedithiocarbazate 

Crystals of the title compound, $\mathrm{C}_{9} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{~S}_{2}$, were obtained from a condensation reaction of methyl dithiocarbazate and benzaldehyde. The planar dithiocarbazate unit is tilted with respect to the phenyl plane with a dihedral angle of 10.96 (12) ${ }^{\circ}$. Intermolecular $\mathrm{N}-\mathrm{H} \cdots \mathrm{S}$ hydrogen bonding stabilizes the crystal structure. A $\mathrm{C}-\mathrm{H} \cdots \pi$ interaction is also observed in the crystal structure.

## Comment

Hydrazine and its derivatives have attracted much attention as they show potential application in the biological field (Okabe et al., 1993; Hu et al., 2001). As part of an ongoing investigation on anticancer compounds, the title compound, (I), has been prepared in our laboratory and its structure is presented here.

(I)

The molecular structure of (I) is shown in Fig. 1. The molecule assumes an $E$ configuration; the phenyl ring and dithiocarbazate unit are located on opposite sides of the $\mathrm{N} 1=\mathrm{C} 7$ bond. The dithiocarbazate unit has a planar configuration and is tilted with respect to the phenyl mean plane, forming a dihedral angle of $10.96(12)^{\circ}$, which agrees well with that of $10.54(8)^{\circ}$ found in a related compound, methyl $\beta-N$-(3nitrophenylmethylene)dithiocarbazate (Zhang et al., 2005). The $\mathrm{C} 7-\mathrm{N} 1-\mathrm{N} 2-\mathrm{C} 8$ torsion angle of $171.69(16)^{\circ}$ shows the relatively poor coplanarity of the whole molecule.

The $\mathrm{C} 8-\mathrm{N} 2$ bond distance (Table 1 ) is much shorter than a typical single $\mathrm{C}-\mathrm{N}$ bond and suggests electron delocalization between the imino and dithiocarboxyl groups.
$\mathrm{N}-\mathrm{H} \cdots \mathrm{S}$ hydrogen bonding is observed between neighboring molecules related by an inversion center (Table 2 and Fig. 1). A $\mathrm{C}-\mathrm{H} \cdots \pi$ interaction occurs in the crystal structure (Fig. 2): $\mathrm{C} 9-\mathrm{H} 9 A \cdots C g=144^{\circ}, \mathrm{H} 9 A \cdots C g=2.87 \AA$ and $\mathrm{C} 9 \cdots C g=3.696(3) \AA$, where $C g$ is the centroid of the $\mathrm{C} 1^{\mathrm{ii}}$ phenyl ring [symmetry code: (ii) $1-x, 1-y, 1-z$ ]. The shortest centroid-to-centroid separation between parallel benzene rings is 4.202 (2) $\AA$, which indicates there is no $\pi-\pi$
$\qquad$


Figure 1
The molecular structure of (I) and a hydrogen-bonded neighbor, with $30 \%$ probability displacement ellipsoids (arbitrary spheres for H atoms). Dashed lines indicate the hydrogen bonds [symmetry code: (i) $1-x,-y, 1$ $-z]$.


Figure 2
The packing, showing the $\mathrm{C}-\mathrm{H} \cdots \pi$ interactions (dashed lines) between neighboring molecules [symmetry code: (ii) $1-x, 1-y, 1-z$ ].
stacking between neighboring molecules in the crystal structure of (I).

## Experimental

Methyl dithiocarbazate was synthesized in the manner reported previously (Hu et al., 2001). Methyl dithiocarbazate ( $1.24 \mathrm{~g}, 10 \mathrm{mmol}$ ) and benzaldehyde ( $1.06 \mathrm{~g}, 10 \mathrm{mmol}$ ) were dissolved in ethanol $(10 \mathrm{ml})$ and refluxed for 4 h . Fine colorless crystals appeared on
cooling. They were separated and washed with cold water three times. Single crystals of (I) were obtained by recrystallization from absolute ethanol.

## Crystal data

$\mathrm{C}_{9} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{~S}_{2}$
$M_{r}=210.31$
Triclinic, $P \overline{1}$
$a=5.909(2) \AA$
$b=9.411(4) \AA$
$c=10.2423(18) \AA$
$\alpha=69.929(11)^{\circ}$
$\beta=80.34(2)^{\circ}$
$\gamma=78.829(15)^{\circ}$
$V=521.7(3) \AA^{\circ}$

$$
\begin{aligned}
& Z=2 \\
& D_{x}=1.339 \mathrm{Mg} \mathrm{~m}^{-3}
\end{aligned}
$$

Mo $K \alpha$ radiation
Cell parameters from 4400
reflections
$\theta=3.5-27.0^{\circ}$
$\mu=0.47 \mathrm{~mm}^{-1}$
$T=295$ (2) K
Prism, colorless
$0.43 \times 0.33 \times 0.30 \mathrm{~mm}$

## Data collection

Rigaku R-AXIS RAPID
diffractometer
$\omega$ scans
Absorption correction: none
5121 measured reflections
2365 independent reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.037$
$w R\left(F^{2}\right)=0.115$
$S=1.08$
2365 reflections
119 parameters

> 1868 reflections with $I>2 \sigma(I)$
> $R_{\text {int }}=0.024$
> $\theta_{\max }=27.5^{\circ}$
> $h=-7 \rightarrow 6$
> $k=-12 \rightarrow 12$
> $l=-13 \rightarrow 13$

H -atom parameters constrained

$$
w=1 /\left[\sigma^{2}\left(F_{\mathrm{o}}^{2}\right)+(0.0722 P)^{2}\right]
$$

where $P=\left(F_{\mathrm{o}}{ }^{2}+2 F_{\mathrm{c}}{ }^{2}\right) / 3$
$(\Delta / \sigma)_{\max }=0.001$ 。
$\Delta \rho_{\text {max }}=0.23 \mathrm{e}^{\circ}{ }^{-3}$
$\Delta \rho_{\text {min }}=-0.33 \mathrm{e}^{-3}$

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| $\mathrm{S} 1-\mathrm{C} 8$ | $1.6630(16)$ | $\mathrm{N} 1-\mathrm{C} 7$ | $1.263(2)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{S} 2-\mathrm{C} 8$ | $1.7442(18)$ | $\mathrm{N} 1-\mathrm{N} 2$ | $1.3755(18)$ |
| $\mathrm{S} 2-\mathrm{C} 9$ | $1.791(2)$ | $\mathrm{N} 2-\mathrm{C} 8$ | $1.331(2)$ |
|  |  |  |  |
| $\mathrm{C} 8-\mathrm{S} 2-\mathrm{C} 9$ | $102.05(9)$ | $\mathrm{N} 2-\mathrm{C} 8-\mathrm{S} 1$ | $120.90(13)$ |
| $\mathrm{C} 7-\mathrm{N} 1-\mathrm{N} 2$ | $115.44(13)$ | $\mathrm{N} 2-\mathrm{C} 8-\mathrm{S} 2$ | $113.69(11)$ |
| $\mathrm{C} 8-\mathrm{N} 2-\mathrm{N} 1$ | $120.56(13)$ | $\mathrm{S} 1-\mathrm{C} 8-\mathrm{S} 2$ | $125.40(11)$ |

Table 2
Hydrogen-bond geometry ( $\AA \mathrm{A}^{\circ}$ ).

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :--- | :--- | :--- | :--- | :--- |
| $\mathrm{~N} 2-\mathrm{H} 2 A \cdots \mathrm{~S}^{\mathrm{i}}$ | 0.86 | 2.59 | $3.431(2)$ | 168 |

Symmetry code: (i) $-x+1,-y,-z+1$.

Methyl H atoms were placed in calculated positions with $\mathrm{C}-\mathrm{H}=$ $0.96 \AA$ and their torsion angles refined to fit the electron density, with $U_{\text {iso }}(\mathrm{H})=1.5 U_{\text {eq }}(\mathrm{C})$. Other H atoms were placed in calculated positions with $\mathrm{C}-\mathrm{H}=0.93$ and $\mathrm{N}-\mathrm{H}=0.86 \AA$, and refined in riding mode with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{C}, \mathrm{N})$

Data collection: PROCESS-AUTO (Rigaku, 1998); cell refinement: PROCESS-AUTO; data reduction: CrystalStructure (Rigaku/ MSC and Rigaku, 2002); program(s) used to solve structure: SIR92 (Altomare et al., 1993); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

## organic papers

The work was supported by the Natural Science Foundation of Zhejiang Province of China (No. M203027) and the Natural Science Foundation of China (No. 20443003).

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